# **CHEMISTRY**

## CEM

# **Department of Chemistry College of Natural Science**

### **Explorations in Chemistry**

Fall, Spring. 3(4-0) P: MTH 103 or concurrently R: Approval of department.

Introduction to core ideas in chemistry (structure and properties of matter, energy, and electrical forces) blended with science practices (use of models, argumentation, construction of scientific explanations, mathematical thinking) to understand and explain chemical phenomena.

#### 141 **General Chemistry**

Fall, Spring, Summer. 4(4-0) P: ((MTH 103 or concurrently) or (MTH 110 or concurrently) or (MTH 116 or concurrently) or (MTH 124 or concurrently) or (MTH 132 or concurrently) or (MTH 152H or concurrently) or (LB 118 or concurrently)) or designated score on Mathematics Placement test

Elements and compounds; reactions; stoichiometry; thermochemistry; atomic structure; chemical bonding; states of matter; solutions; acids and bases; aqueous equilibria.

### **General and Inorganic Chemistry**

Fall, Spring, Summer. 3(4-0) P: CEM 141 or CEM 151 or CEM 181H or LB 171

Kinetics; gaseous equilibria; acids and bases; pH; buffers; hydrolysis; titrations; heterogeneous equilibria; thermodynamics; redox and electrochemistry; transition metal chemistry; nuclear chemistry; main group chemistry.

### Survey of Organic Chemistry 143

Fall, Spring, Summer. 4(3-3) P: CEM 141 or CEM 151 or CEM 181H or LB 171 Not open to students with credit in CEM 351.

Chemistry of carbon compounds. Chemistry of the main organic functional groups with applications to everyday life, industry, and biology.

#### 151 **General and Descriptive Chemistry**

Fall. 4(4-0) P: ((MTH 116 or concurrently) or (MTH 124 or concurrently) or (MTH 132 or concurrently) or (MTH 152H or concurrently) or (LB 118 or concurrently)) or designated score on Mathematics Placement test

Stoichiometry; solutions; reactions and thermochemistry; quantum mechanics and atomic structure; periodic properties; chemical bonding; molecular structure; coordination chemistry; organic molecules and functional groups.

### 152

**Principles of Chemistry** Spring. 3(4-0) P: CEM 151 or CEM 181H or LB 171

Gases, liquids, and solids; thermodynamics; changes of state; solutions and colligative properties; chemical equilibria; acids, bases, and aqueous equilibria; kinetics; redox reactions and electrochemistry; nuclear chemistry.

#### 161 Chemistry Laboratory I

Fall, Spring, Summer. 1(0-3) P: (CEM 141 or concurrently) or (CEM 151 or concurrently) or (CEM 181H or concurrently) or (LB 171 or concurrently)

Experiments in general chemistry; stoichiometry, calorimetry, electrochemistry, molecular geometry, gas laws, kinetics, acids and bases, and inorganic chemistry.

#### 162 **Chemistry Laboratory II**

Fall, Spring, Summer. 1(0-3) P: CEM 161 or CEM 185H or LB 171L RB: (CEM 142 or concurrently) or (CEM 152 or concurrently) or (CEM 182H or concurrently)

Analytical and inorganic chemistry; redox and acid base titrations; spectrophotometric and gravimetric analysis; preparation and analysis of coordination complexes of nickel, iron, and cobalt.

### Honors Chemistry I

Fall. 4(4-0) P: (MTH 124 or concurrently) or (MTH 132 or concurrently) or (MTH 152H or concurrently) or (LB 118 or concurrently) R: Approval of department.

Atomic structure and quantum mechanics; chemical bonding and molecular structure; spectroscopy; coordination chemistry; materials or biological macromolecules.

### Honors Chemistry II

Spring. 4(4-0) P: (ČEM 151 or CEM 181H or LB 171) and ((MTH 126 or concurrently) or (MTH 133 or concurrently) or (MTH 153H or concurrently) or (LB 119 or concurrently)) R: Approval of department.

Gases, solids, liquids, solutions, and phase transitions; thermodynamics; spontaneity and the second law of thermodynamics; chemical equilibrium; acidbase equilibria; redox reactions and electrochemistry; kinetics.

### 185H

Honors Chemistry Laboratory I Fall. 2(1-3) P: CEM 181H or concurrently R: Approval of department.

Spectroscopic methods used to determine the struc-

ture of molecules and materials. Experiments applying principles of physical, organic, inorganic, analytical, biological, and materials chemistry, while introducing analytical (qualitative and quantitative) and synthetic techniques.

## Organic Chemistry I

Fall, Spring, Summer. 3(4-0) P: CEM 141 or CEM 151 or CEM 181H or LB 171 Not open to students with credit in CEM 351.

Common classes of organic compounds including their nomenclature, structure, bonding, reactivity, and spectroscopic characterization.

## Organic Chemistry II

Fall, Spring, Summer. 3(4-0) P: CEM 251 Not open to students with credit in CEM

Continuation of CEM 251 with emphasis on polyfunctional compounds, particularly those of biological interest

#### 255 **Organic Chemistry Laboratory**

Fall, Spring, Summer. 2(1-3) P: (CEM 252 or concurrently) and (CEM 161 or LB 171L or CEM 185H) Not open to students with credit in CEM 355.

Preparation and qualitative analysis of organic compounds.

#### **Quantitative Analysis** 262

Fall, Spring, Summer. 3(3-3) P: (CEM 142 or CEM 152 or CEM 182H or LB 172) and (CEM 162 or CEM 185H or LB 172L)

Introduction to analytical chemistry and quantitative methods; aqueous solution equilibria and statistics related to quantitative chemical analysis; titrimetric, gravimetric, and spectrophotometric measurements.

#### 311 **Inorganic Chemistry**

Fall. 3(3-0) P: CEM 142 or CEM 152 or CEM 182H or LB 172 RB: CEM 384

Basic symmetry, molecular orbital theory, and valence bond theory applications to inorganic systems. Physical properties and reactivity of transition metal

### Instrumental Methods and Applications

Spring. 3(2-3) P: {CEM 262 or (CEM 162 and BLD 213 and BLD 417)} and ((CEM 143 or CEM 251 or CEM 351) and completion of

Tier I writing requirement)

Principles and applications of instrumental analysis of separation techniques.

Organic Chemistry I
Fall. 3(4-0) P: CEM 152 or CEM 182H or
CEM 142 or LB 172 Not open to students with credit in CEM 251.

Structure, bonding, and reactivity of organic mole-

#### 352 Organic Chemistry II

Spring. 3(4-0) P: CEM 351 Not open to students with credit in CEM 252.

Carboxylate derivatives. Conjugation. Aromaticity. acids. Proteins. Carbohydrates. Nucleic Amino

#### Organic Laboratory I 355

Spring. 2(0-6) P: (CEM 162 or CEM 185H or LB 172L) and (((CEM 352 or concurrently) or (CEM 252 or concurrently)) and completion of Tier I writing requirement)

Organic laboratory techniques. Distillation. Spectroscopy. Melting points. Recrystallization. Chromatography. Measuring physical properties.

## Organic Laboratory II

Fall. 2(0-6) P: CEM 355

Multi-step organic synthesis. Qualitative organic analysis. Separation, identification, and characterization of unknowns.

# Introductory Physical Chemistry I

Fall. 3(4-0) P: (CEM 142 or CEM 152 or CEM 182H or LB 172) and (MTH 133 or MTH 153H or MTH 126 or LB 119) RB: PHY 184 or PHY 232 or PHY 232C or PHY 294H or LB 274 SA: CEM 391

Physical chemistry of macroscopic systems: thermodynamics, kinetics, electrochemistry.

#### Introductory Physical Chemistry II 384

Spring. 3(4-0) P: (CEM 142 or CEM 152 or CEM 182H or LB 172) and (MTH 133 or MTH 153H or MTH 126 or LB 119) and (PHY 184 or PHY 232 or PHY 232C or PHY 294H or LB 274) RB: CEM 383

Physical chemistry of microscopic systems: quantum mechanics, spectroscopy.

# Analytical/Physical Laboratory

Spring. 2(1-4) P: (CEM 483 and (CEM 484 or concurrently) and CEM 262) and completion of Tier I writing requirement SA: CEM 372, CEM 472

Chemical kinetics, thermodynamics, and computerbased data analysis methods.

#### 400H **Honors Work**

Fall, Spring, Summer. 1 to 12 credits. A student may earn a maximum of 12 credits in all enrollments for this course. P: Completion of Tier I writing requirement. R: Approval of department.

Readings and investigations in chemistry.

#### 411 **Advanced Inorganic Chemistry**

Spring. 4(4-0) P: CEM 311 or CEM 384 or **CEM 483** 

Principles of structure and bonding. Symmetry. Solid state chemistry. Acid-base and redox reactions. Main group chemistry: transition metal bonding, spectra, and reaction mechanisms.

### **Advanced Synthesis Laboratory**

Spring. 3(0-8) P: (CEM 411 and CEM 356) and completion of Tier I writing requirement RB: CEM 495 R: Open to juniors or seniors in the Bachelor of Science in Chemistry or in the Lyman Briggs Chemistry Coordinate Major or approval of department.

Methods of synthesizing inorganic and organometallic compounds.

#### 419 Independent Study

Fall, Spring, Summer. 1 to 12 credits. A student may earn a maximum of 12 credits in all enrollments for this course. P: Completion of Tier I Writing Requirement R: Approval of department.

Faculty supervised readings in chemistry.

### Independent Research

Fall, Spring, Summer. 1 to 12 credits. A student may earn a maximum of 12 credits in all enrollments for this course. RB: Completion of Tier I Writing requirement R: Approval of department.

Faculty supervised independent investigations in chemistry.

### 425 **Chemistry Communication and**

Professional Development (W)
Fall. 3(3-0) P: ((CEM 262) and completion of Tier I writing requirement) and (CEM 255 or CEM 355) R: Open to students in the Chemistry Major or approval of department.

Written and oral communication skills for entering and participating in the chemistry profession and post-undergraduate programs. Includes discussion of academic honesty and research integrity.

#### **Advanced Analytical Chemistry** 434

Fall. 3(3-1) P: CEM 395 and CEM 352 and CEM 484 SA: CEM 361, CEM 362

Instrumental methods of analysis, including spectroscopy, chromatography, and electrochemistry.

#### 435 **Analytical Chemistry Laboratory**

Spring. 3(1-4) P: (CEM 434) and completion of Tier I writing requirement SA: CEM 372, CFM 472

Application of instrumental spectroscopic, electrochemical, and chromatographic methods to solve quantitative chemical problems in the laboratory.

# **Chemical Safety**

Fall. 1(1-0) P: (CEM 142 or CEM 152 or CEM 182H or LB 172) and (CEM 252 or CEM 352)

Prudent laboratory practices. Regulatory agencies' expectations of chemical industries and academia.

#### 481 Seminar in Computational Chemistry

Fall of odd years. 3(2-3) P: (CEM 384 or CEM 483 or PHY 471) and MTH 235 RB: MTH 309 or MTH 314 or MTH 317H

Potential energy surfaces; matrix representation of quantum mechanics; linear combination of atomic orbitals; Hartree-Fock approximation; electron correconfiguration interaction; coupled cluster theory; Moller Plesset perturbation theory; density functional theory

#### 482 Science and Technology of Wine Production

Fall. 3(2-3) Interdepartmental with Chemical Engineering and Food Science. Administered by Chemistry. P: CEM 143 or CEM 251 or CEM 351 RB: Must be at least 21 years of age. R: Open to seniors or graduate students in the Department of Biosystems and Agricultural Engineering or in the Department of Chemical Engineering and Materials Science or in the Department of Chemistry or in the Department of Food Science and Human Nutrition or in the Department of Horticulture or in the Department of Microbiology and Molecular Genetics or in the Lyman Briggs Chemistry Coordinate Major. Approval of department.

Origin and history of wine and wine production.

Determination and timing of harvest, methods of postharvest handling, storage, and processing of grapes into juice and wine. Physical and chemical changes in wine and processes. Analysis of must and its adjustment, fermentation, fining, and aging. Physiology of yeasts and bacteria involved in winemaking and spoilage. Cellar practices, problems, and operations.

#### 483 **Quantum Chemistry**

Fall. 3(4-0) P: (MTH 235 or MTH 255H or MTH 347H or MTH 340) and (PHY 184 or PHY 294H or LB 274 or PHY 184B) and (CEM 142 or CEM 152 or CEM 181H or LB 172) SA: CEM 362, CEM 461

Postulates of quantum mechanics and the application to model systems, atoms and molecules. Introduction to molecular spectroscopy.

#### 484 **Molecular Thermodynamics**

Spring. 3(4-0) P: (MTH 235 or MTH 255H or MTH 340 or MTH 347H) and (CEM 142 or CEM 152 or CEM 182H or LB 172) RB: CEM 483 SA: CEM 361, CEM 391

Microscopic properties of atoms and molecules revealed by spectroscopic measurements; connection between thermodynamic properties of macroscopic chemical systems and microscopic properties established using statistical thermodynamics.

#### 485 Modern Nuclear Chemistry

Spring of even years. 3(3-0) P: (CEM 142 or CEM 152 or CEM 182H or LB 172) and (PHY 184 or PHY 232 or PHY 294H or LB 274) RB: CEM 483 or CEM 384 or PHY 471 SA: CEM 430

Elementary nuclear processes and properties; radioactivity, its measurement and its interaction with matter.

## **Molecular Spectroscopy**

Fall. 2(1-4) P: (CEM 483 or CEM 484) and (CEM 395 or CEM 499) and ((CEM 262 or CEM 186H) and completion of Tier I writing requirement) SA: CEM 472

Experiments in magnetic resonance, optical, and vibrational spectroscopies.

#### 499 **Chemical Physics Seminar**

Spring. 1(1-0) A student may earn a maximum of 2 credits in all enrollments for this course. P: ((PHY 215) and completion of Tier I writing requirement) and (MTH 235 or MTH 255H or MTH 340 or MTH 347H)

Written and oral reports on selected journal articles in chemical physics.

#### 811 Advanced Inorganic Chemistry I

Fall. 3(3-0) R: Open only to graduate students in College of Natural Science or College of Engineering.

Principles of chemical bonding, electronic structure, and reaction mechanisms of main group and transition metal compounds. Concepts of group theory.

### Advanced Inorganic Chemistry II

Spring. 3(3-0) RB: CEM 811 R: Open only to graduate students in College of Natural Science or College of Engineering.

Descriptive chemistry of inorganic compounds. Emphasis on synthesis, structure, and reactivity patterns of coordination, organometallic, and solid state compounds of transition metals and main group elements.

### Organometallic Chemistry 820

Fall. 3(3-0)

Organometallic functional groups. Principles of electronic structure, and bonding in organometallic species will be related to reactivity patterns in common systems. Preparation of complexes with applications to catalytic and stoichiometric organic syntheses.

#### Mass Spectrometry 832

Fall, Spring. 3(3-0) R: Open only to graduate students in the College of Natural Science or College of Engineering.

Instrumentation of mass spectrometry. Interpreting mass spectra of organic and inorganic molecules. Applications to analysis of large molecules and chromatography.

#### 834 Advanced Analytical Chemistry I

Fall. 3(3-0) R: Open to graduate students in the College of Engineering or in the College of Natural Science.

Basic electronics and data acquisition and anaylsis, electrochemistry, and statistics for chemists.

## **Advanced Analytical Chemistry II**

Fall. 3(3-0) R: Open to graduate students in the College of Engineering or in the College of Natural Science or in the School of Criminal Justice.

Separations, molecular spectroscopy and mass spectrometry

#### 836 Separation Science

Spring of odd years. 3(3-0) R: Open only to graduate students in College of Natural Science or College of Engineering.

Physical and chemical principles of separations, column technology, and instrumentation for gas, liquid, and supercritical fluid chromatography.

### 837

**Electroanalytical Chemistry**Fall of even years. 3(3-0) R: Open only to graduate students in College of Natural Science or College of Engineering.

Modern electroanalytical chemistry. Theory and applications to chemical and biological problems. Coulometry, voltammetry, ion-selective potentiometry, and other electrochemical techniques.

### Structure and Spectroscopy of Organic Compounds

Fall. 3(3-0) R: Open only to graduate students in College of Natural Science or College of Engineering.

Structural and stereochemical principles in organic chemistry. Applications of spectroscopic methods, especially nuclear magnetic resonance, static and dynamic aspects of stereochemistry. Spectroscopy in structure determination.

#### 850 Intermediate Organic Chemistry

Fall. 3(3-0)

Traditional and modern basic reaction mechanisms and principles and their synthetic applications.

### **Advanced Organic Chemistry**

Fall. 3(3-0) R: Open only to graduate students in College of Natural Science or Col-

lege of Engineering.
Structure, reactivity, and methods. Acid-base reactions, substitution, addition, elimination, and pericyclic processes. Major organic intermediates related to simple bonding theory, kinetics, and thermodynamics

#### 852 **Methods of Organic Synthesis**

Spring. 3(3-0) R: Open only to graduate students in College of Natural Science or College of Engineering.

Principal reactions leading to carbon-carbon bond formation and functional group transformations. Strategies and methods of organic synthesis.

#### 881 **Atomic and Molecular Structure**

Fall. 3(3-0) R: Open only to graduate students in College of Natural Science or College of Engineering.

Postulates of quantum mechanics, analytical solutions of the Schroedinger equation, theoretical descriptions of chemical bonding, spectroscopy, statistical mechanics, and statistical thermodynamics.

#### 882 **Kinetics and Spectroscopic Methods**

Spring. 3(3-0) R: Open only to graduate students in College of Natural Science or College of Engineering.

Rate equations and mechanisms of chemical reactions: reaction rate theory, kinetic theory of gases, photochemistry. Spectroscopic methods, and applications of spectroscopy in reaction kinetics.

#### **Computational Quantum Chemistry** 883

Fall. 3(2-3) RB: CEM 461 or CEM 881 Computational methods in determining electronic

energy levels, equilibrium nuclear configurations, and other molecular properties.

# **Computational Chemistry**

Spring. 3(2-3)

Computational approaches to molecular problems. Use of ab initio and semi-empirical electronic structure, molecular mechanics and molecular dynamics

#### **Chemical Problems and Reports** 890

Fall, Spring, Summer. 1 to 6 credits. A student may earn a maximum of 12 credits in all enrollments for this course.

Investigation and report of a nonthesis problem in chemistry.

#### 899 Master's Thesis Research

Fall, Spring, Summer. 1 to 20 credits. A student may earn a maximum of 99 credits in all enrollments for this course. R: Open only to graduate students in the Department of Chemistry.

Master's thesis research.

### **Selected Topics in Inorganic Chemistry**

Fall, Spring. 1 to 3 credits. A student may earn a maximum of 9 credits in all enrollments for this course. R: Open to graduate students in the Department of Chemistry or approval of department.

Current research topics in inorganic chemistry.

#### 918 **Inorganic Chemistry Seminar**

Fall, Spring. 1(1-0) A student may earn a maximum of 3 credits in all enrollments for this course. R: Open to graduate students in the Department of Chemistry.

Advances in inorganic chemistry reported by gradu-

### Selected Topics in Analytical Chemistry

Fall, Spring, 2 to 3 credits. A student may earn a maximum of 9 credits in all enrollments for this course. R: Open only to graduate students in College of Natural Science or College of Engineering.

Advanced computer techniques, surface chemistry, analytical chemistry of polymers, or statistics for

#### 938 **Analytical Chemistry Seminar**

Fall, Spring. 1(1-0) A student may earn a maximum of 3 credits in all enrollments for this course. R: Open to graduate students in the College of Engineering or in the College of Natural Science.

Advances in analytical chemistry reported by graduate students, faculty, and guest lecturers.

### Selected Topics in Organic Chemistry

Fall, Spring. 1 to 3 credits. A student may earn a maximum of 12 credits in all enrollments for this course. R: Open only to graduate students in College of Natural Science or College of Engineering.

Heterocyclic and organometallic chemistry, natural

products, photochemistry, free radicals, or reaction mechanisms.

#### **Organic Chemistry Seminar** 958

Fall, Spring. 1(1-0) A student may earn a maximum of 2 credits in all enrollments for this course. R: Open to graduate students in the College of Engineering or in the College of Natural Science.

Advances in organic chemistry reported by graduate students.

### **Emerging Topics in Chemistry**

Fall, Spring. 1 to 3 credits. A student may earn a maximum of 6 credits in all enrollments for this course. R: Open only to doctoral students in the Chemistry or Chemical Physics major.

Discussion of a research topic of emerging interest in chemistry. Preparation of a proposal for funding of research.

#### 985 **Selected Topics in Nuclear Chemistry**

Fall. 1 to 3 credits. A student may earn a maximum of 12 credits in all enrollments for this course. RB: Thermodynamics, Statistical Mechanics, Quantum Mechanics, Electricity and Magnetism, Differential and Integral Calculus, Differential Equations R: Open to doctoral students in the College of Engineering or in the College of Natural Science or in the Department of Chemistry.

Nuclear instruments, detectors and electronics, vacuum technology, electric and magnetic properties of nuclei, nuclear simulation tools, or nuclear spectroscopy and reactions.

#### Selected Topics in Physical Chemistry I 987

Fall. 1 to 3 credits. A student may earn a maximum of 9 credits in all enrollments for this course. R: Open only to doctoral students or approval of department.

Topics such as kinetics and photochemistry, macromolecular and surface chemistry, molecular spectroscopy, electric and magnetic properties of matter, or applications of statistical mechanics to chemical

### 988

Selected Topics in Physical Chemistry II Spring. 1 to 3 credits. A student may earn a maximum of 9 credits in all enrollments for this course. R: Open only to doctoral students or approval of department.

Topics such as analysis and interpretation of molecular spectra, advanced molecular structure theory, magnetic resonance, X-rays and crystal structure, scientific analysis of vacuum systems, or problems in statistical mechanics.

#### 991 **Quantum Chemistry and Statistical** Thermodynamics I

Fall. 3(3-0) R: Open only to graduate students in College of Natural Science or College of Engineering.

Principles and applications of quantum chemistry. Partition functions, spectroscopic measurements, and thermodynamic applications.

### **Quantum Chemistry and Statistical** 992 Thermodynamics II

Spring. 3(3-0) RB: CEM 991

Analytical and numerical methods for solving quantum chemical problems. Statistical mechanics of solids and liquids.

## **Advanced Topics in Quantum Chemistry**

Spring. 3(3-0) A student may earn a maximum of 9 credits in all enrollments for this course. R: Open to graduate students in the College of Engineering or in the College of Natural Science.

Spectroscopic theory, properties of atoms and molecules in electric and magnetic fields, intermolecular forces. Many-body theory, molecular electronic structure, solid state chemistry, or molecular reaction dynamics.

#### **Advanced Topics in Statistical** 994 Mechanics

Fall. 3(3-0) A student may earn a maximum of 9 credits in all enrollments for this course. R: Open to graduate students in the College of Engineering or in the College of Natural Science.

Nonequilibrium statistical mechanics and thermodynamics. Correlation functions and spectroscopy, light scattering, magnetic relaxation, transport properties of fluids and gases, or statistical mechanics of chemical reactions.

### **Nuclear Chemistry Seminar**

Fall, Spring. 1 credit. A student may earn a maximum of 2 credits in all enrollments for this course. RB: One year of graduate work in nuclear chemistry or related experience R: Open to graduate students in the Department of Chemistry or in the Department of Physics and Astronomy.

Advances in nuclear chemistry reported by graduate students, faculty, and guest lecturers.

# Chemistry—CEM

998 Physical Chemistry Seminar
Fall, Spring. 1(1-0) A student may earn a maximum of 3 credits in all enrollments for this course. R: Open to graduate students in the Department of Chemistry.

Advances in physical chemistry reported by graduate students.

999 Doctoral Dissertation Research
Fall, Spring, Summer. 1 to 24 credits. A
student may earn a maximum of 36 credits
in all enrollments for this course. R: Open to
doctoral students in the Department of
Chemistry.
Doctoral dissertation research.